

# A short access to highly strained spiranic compounds from ethyl 3-cyclobutylprop-2-enoate

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Received 12 February 2008; accepted 29 February 2008  
Available online 6 March 2008

## Abstract

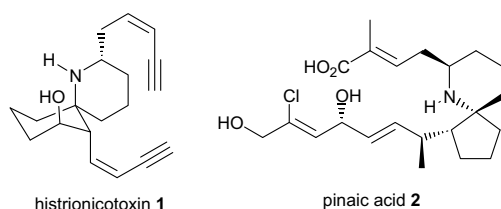
UV irradiation of ethyl 3-cyclobutylprop-2-enoate delivered the  $\beta,\gamma$ -unsaturated isomer in high yield. Its C=C double bond was submitted to epoxidation and cyclopropanation to deliver the corresponding spiro[3.2]hexane derivatives. Alternatively, the same substrate treated by TMS-I or by OsO<sub>4</sub> allowed an easy access to two spiranic butyrolactones. © 2008 Elsevier Ltd. All rights reserved.

**Keywords:** Photochemistry; Isomerization; Cyclopropanation; Epoxidation; Lactone

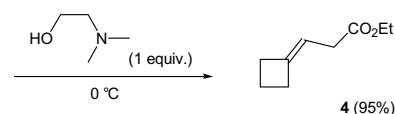
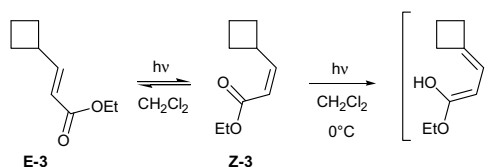
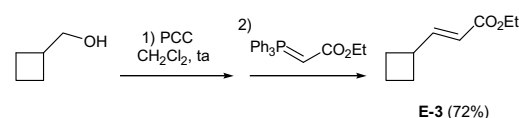
Spiranic compounds due to their atypical topologies have always attracted the attention of synthetic chemists.<sup>1</sup> It is noteworthy that some of them possess important biological activities like histrionicotoxin **1** which exhibits neuro-physiological properties<sup>2</sup> or pinnaic acid **2** which possess anti-inflammatory properties<sup>3,4</sup> (Scheme 1). In contrast, more strained structures in which the spiranic centre is connected either to a three or to a four-membered ring are rarely found in Nature. Nevertheless, their synthesis has been considered in the context of energy storage<sup>5,6</sup> or as prodrugs.<sup>7,8</sup>

In connection with our interest into photochemical processes, we already reported the direct formation of spiranic

$\beta$ -lactams by irradiation of 3-carboxamido-cyclohex-2-enones.<sup>9</sup> Alternatively, larger spiranic lactones and lactams were conveniently obtained by the selective cleavage of cyclobutane derivatives.<sup>10</sup> Here, we wish to describe the synthesis of heterocyclic spirocompounds possessing a cyclobutane ring directly connected to a second 3 or 5-membered ring. Our strategy is based on the functionalization of ethyl

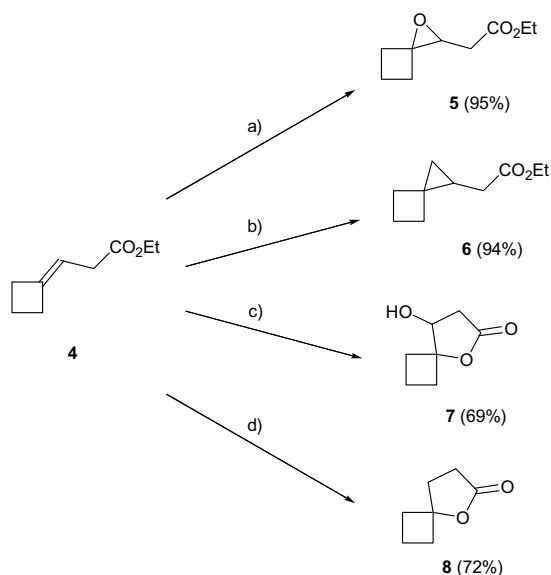


Scheme 1.



Scheme 2.

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Scheme 3. Reagents and conditions: (a) *m*-CPBA, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C; (b) (i) Et<sub>2</sub>Zn, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 10 min; (ii) CH<sub>2</sub>I<sub>2</sub>, 0 °C then rt; (c) OsO<sub>4</sub>, NMO, acetone/H<sub>2</sub>O (1:1), t.a., overnight; (d) TMS–Cl, NaI, CH<sub>3</sub>CN, reflux.

3-cyclobutenylidene-3-propenoate **4** which could be obtained by photoisomerization of  $\alpha,\beta$ -unsaturated ester **3**<sup>11,12</sup> (Scheme 2).

Irradiation of *E*-ethyl 3-cyclobutylpropen-2-olate **3** at 254 nm in dichloromethane and in the presence of *N,N*-dimethylaminoethanol afforded the corresponding  $\beta,\gamma$ -unsaturated ester **4** in a very high yield. It is worth to note that the presence of the  $\beta$ -aminoalcohol is crucial to promote the protonation of the photodiene intermediate.

Compound **4** was first submitted to the action of *m*-CPBA to give in 95% yield the corresponding ethyl (6-[5-oxaspiro[3,2]hexanyl)acetate **5** which was plenty characterized according to <sup>1</sup>H NMR and MS spectroscopy.<sup>13</sup> Similarly, cyclopropanation of **4** was conveniently achieved in a similar yield by using diiodomethane and diethylzinc according to a well-established procedure<sup>14</sup> (Scheme 3).

The synthesis of spiro[4,3] octane derivatives was also investigated. Dihydroxylation of **4** furnished directly the parent  $\beta$ -hydroxylactone **7** in 69% yield. Action of TMS–I<sup>15,16</sup> easily generated in situ delivered in 72% yield, the known butyrolactone **8**.<sup>17</sup>

In conclusion, we have demonstrated the easy conversion of ethyl 3-cyclobutylidenepropanoate into different highly strained spiranic compounds which could be of interest as building blocks in organic synthesis.

## Acknowledgements

We thank the CNRS and Région Rhône-Alpes (Programme Cible 2007) for financial support. H.S. thanks the Syrian Government for a Ph.D. grant.

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- Selective data for new isolated compounds. Compound **4**: <sup>1</sup>H NMR (300 MHz): 1.25 (3H, t, *J* = 7.1 Hz), 1.85–2.05 (2H, m), 2.58–2.73 (4H, m), 2.89 (2H, d, *J* = 6.0 Hz), 4.21 (2H, q, *J* = 7.1 Hz), 5.15–5.25 (1H, m). <sup>13</sup>C NMR (75 MHz): 14.4, 17.0, 29.4, 31.1, 33.9, 60.7, 112.2, 144.5, 172.7. Compound **5**: <sup>1</sup>H NMR (300 MHz): 1.25 (3H, t, *J* = 7.1 Hz), 1.55–2.61 (8H, m), 3.19 (1H, t, *J* = 6.0 Hz), 4.19 (2H, q, *J* = 7.1 Hz), 1.55–2.61 (8H, m), 3.19 (1H, t, *J* = 6.0 Hz), 4.19 (2H, q, *J* = 7.1 Hz). <sup>13</sup>C NMR (75 MHz): 13.1, 14.4, 28.6, 31.1, 36.3, 57.1, 61.1, 63.7, 170.7. IR: 2934, 2857, 1738, 1489, 1448, 1369, 1174, 823 cm<sup>-1</sup>. HRMS: calcd for C<sub>9</sub>H<sub>14</sub>O<sub>3</sub>, 170.8634; found, 170.8639. Compound **6**: <sup>1</sup>H NMR (300 MHz): 0.13 (1H, dd, *J* = 1.3 and 5.1 Hz), 0.63 (1H, ddd, *J* = 1.7, 5.1 and 8.5 Hz), 0.84–0.97 (1H, m), 1.25 (3H, t, *J* = 7.2 Hz), 1.90–2.05 (6H, m), 2.14 (2H, dd, *J* = 1.7 and 7.1 Hz); 4.13 (2H, q, *J* = 7.1 Hz). <sup>13</sup>C NMR (75 MHz): 14.5, 17.5, 18.2, 18.3, 23.7, 26.6, 30.2, 36.2, 60.5, 173.7. IR: 2929, 2857, 1738, 1492, 1448, 1369, 1176, 820 cm<sup>-1</sup>. HRMS: calcd for C<sub>10</sub>H<sub>16</sub>O<sub>2</sub>, 168.1150; found, 168.1145. Compound **7**: <sup>1</sup>H NMR (300 MHz): 1.55–2.17 (5H, m), 2.27–2.60 (m, 4H), 2.77 (1H, dd, *J* = 5.2 and 18.0 Hz), 4.39 (1H, dd, *J* = 1.7 and 5.3 Hz). <sup>13</sup>C NMR (75 MHz): 12.2, 27.8, 32.7, 37.8, 72.4, 88.6, 176.8. IR: 3434, 2929, 2942, 1765, 1494, 1403, 1300, 1055, 875 cm<sup>-1</sup>. HRMS: calcd for C<sub>7</sub>H<sub>10</sub>O<sub>3</sub>, 142.0630; found, 142.0630. Compound **8**: <sup>1</sup>H NMR (300 MHz): 1.60–2.10 (6H, m), 2.27 (2H, t, *J* = 8.0 Hz), 2.52 (2H, t, *J* = 8.0 Hz). <sup>13</sup>C NMR (75 MHz): 12.3, 28.9, 33.7, 34.8, 85.4, 176.8. IR: 2986, 2942, 2880, 1775, 1422, 1292, 1182, 1146, 1107, 955, 909 cm<sup>-1</sup>.
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